

Assignment 6

Discussed during the tutorial on Thursday, December 1st, 2022

- 6.1 (10 points) Consider the system Ax = b where the rows of A are linearly independent. Let \bar{x} be a solution to Ax = b. Let J be the set of column indices j of A for which $\bar{x}_j \neq 0$.
 - (a) Show that if \bar{x} is a basic solution, then the columns of A_J are linearly independent.
 - (b) Show that if the columns of A_J are linearly independent, then \bar{x} is a basic solution for some basis $B \supseteq J$.

Note that (a) and (b) give you a way of checking whether \bar{x} is a basic solution, namely you simply need to verify whether the columns of A_J are linearly independent.

(c) Consider the system of equations

and the following vectors:

- (i) $(1, 1, 0, 0, 0, 0, 0)^T$
- (ii) $(2, -1, 2, 0, 1, 0, 0)^T$
- (iii) $(1, 0, 1, 0, 1, 0, 0)^T$
- (iv) $(0, 0, 1, 1, 0, 0, 0)^T$
- (v) $(0, 1/2, 0, 0, 1/2, 0, 1)^T$

For each vector in (i)-(v), indicate if it is a basic solution or not. Please justify your answers.

- (d) Which of the vectors in (i)-(v) are basic feasible solutions?
- 6.2 (10 points) Consider the LP max{ $c^T x \mid Ax = b, x \ge 0$ } where

$$A = \begin{pmatrix} 1 & 2 & -2 & 0 \\ 0 & 1 & 3 & 1 \end{pmatrix}, b = \begin{pmatrix} 2 \\ 5 \end{pmatrix}, c = \begin{pmatrix} 0 \\ 3 \\ 1 \\ 0 \end{pmatrix}$$

- (a) Beginning with the basis $B = \{1, 4\}$, solve the problem. At each step, choose the entering variable and leaving variable by Bland's rule.
- (b) Give a certificate of optimality or unboundedness for the problem, and verify it.
- 6.3 (10 points) The princess' wedding ring can be made from four types of gold 1,2,3,4 with the following amounts of milligrams of impurity per gram:

Type	1	2	3	4
mg of lead	1	2	2	1
mg of cobalt	0	1	1	2
value	1	2	3	2

Set up an LP which will determine the most valuable ring that can be made containing at most 6 mg of lead and at most 10 mg of cobalt. Put the LP into SEF and then solve it.

Upload your solutions as a $\,.\mathtt{pdf}\text{-}\mathrm{file}$ to the course page on the TUHH e-learning portal until 8am on Tuesday, November 29th.